

Accident Tolerant Control Rod Development in Japan

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Fuel & CR behavior in early stage of SA

Fuel Pin: H_2O / Zircaloy / UO_2

Oxidation reaction of Zry by high-temperature steam accelerates intensely at $>1200^\circ\text{C}$.



Fuel temperature increases sharply.



Zry cladding melts at $\sim 1760^\circ\text{C}$.



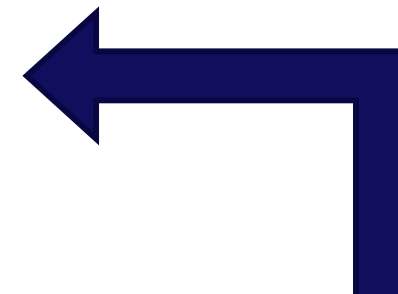
Fuel material, UO_2 , dissolves in molten Zry.

Control Rod: Zircaloy, H_2O / S.S. / B_4C or Ag-In-Cd

Eutectic temperature of Fe / Zr = 940°C

Eutectic temperature of B_4C / Fe = 1150°C

Melting point of Ag-In-Cd = 800°C



In early stage of severe accident, CR potentially melt and fall down into lower part of core prior to fuel melt



Novel Control Rod Concept

Accident Tolerant Control Rod:

- to attain sub-criticality at any accidental conditions including SA.

- ① As sufficient neutron absorption effect as conventional CR materials.
- ② Higher melting & eutectic temperature than $\sim 1200^{\circ}\text{C}$. Or
- ②' Higher resistance to eutectic reaction
- ③ High miscibility with molten and solidified fuel materials.

Candidate Materials

Absorber: RE_2O_3 (Sm_2O_3 , Gd_2O_3 , Eu_2O_3 ), HfO_2 , etc. (①, ②, ③)

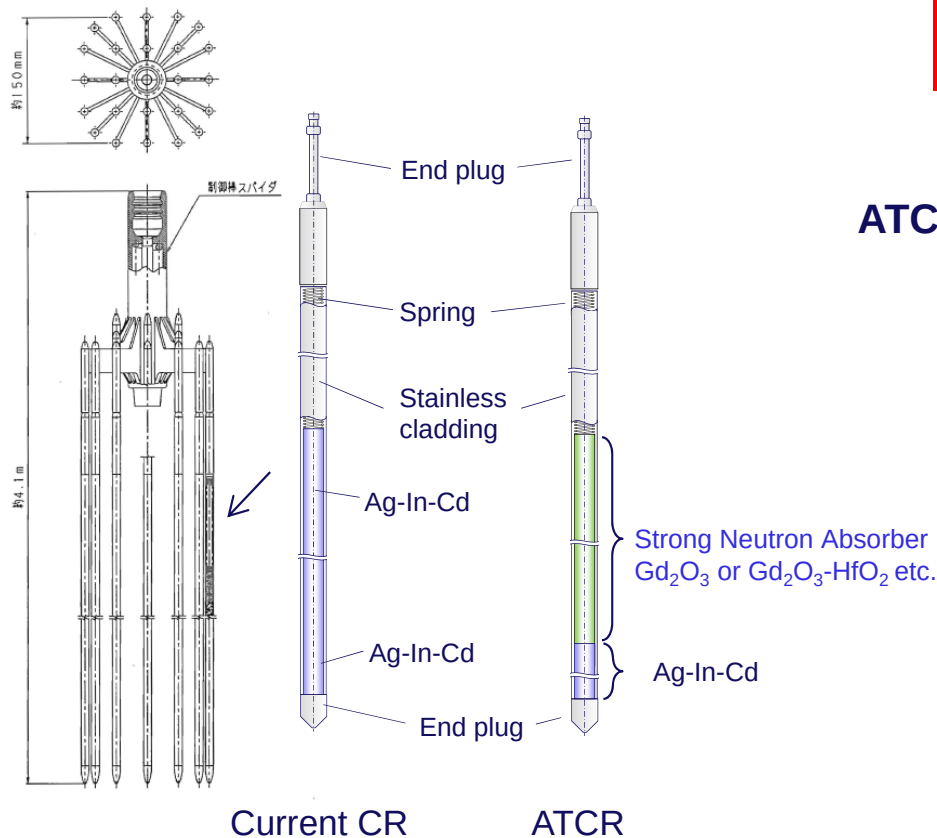
Structure: Fe-Cr-Al ODS ferritic steel, etc. (①, ②')

- to be applicable to current LWRs.

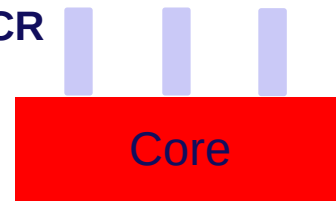
- ④ Influence on normal operation is minimized.
- ⑤ CR worth is kept for a long term.

Operation concept of ATCR – Absorber Modification –

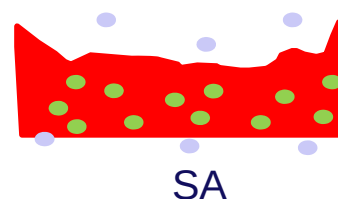
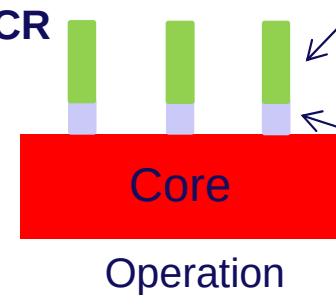
Example in PWR



Current CR



ATCR



- Ag-In-Cd
- ATCR material
RE₂O₃ or RE₂O₃-HfO₂ etc.

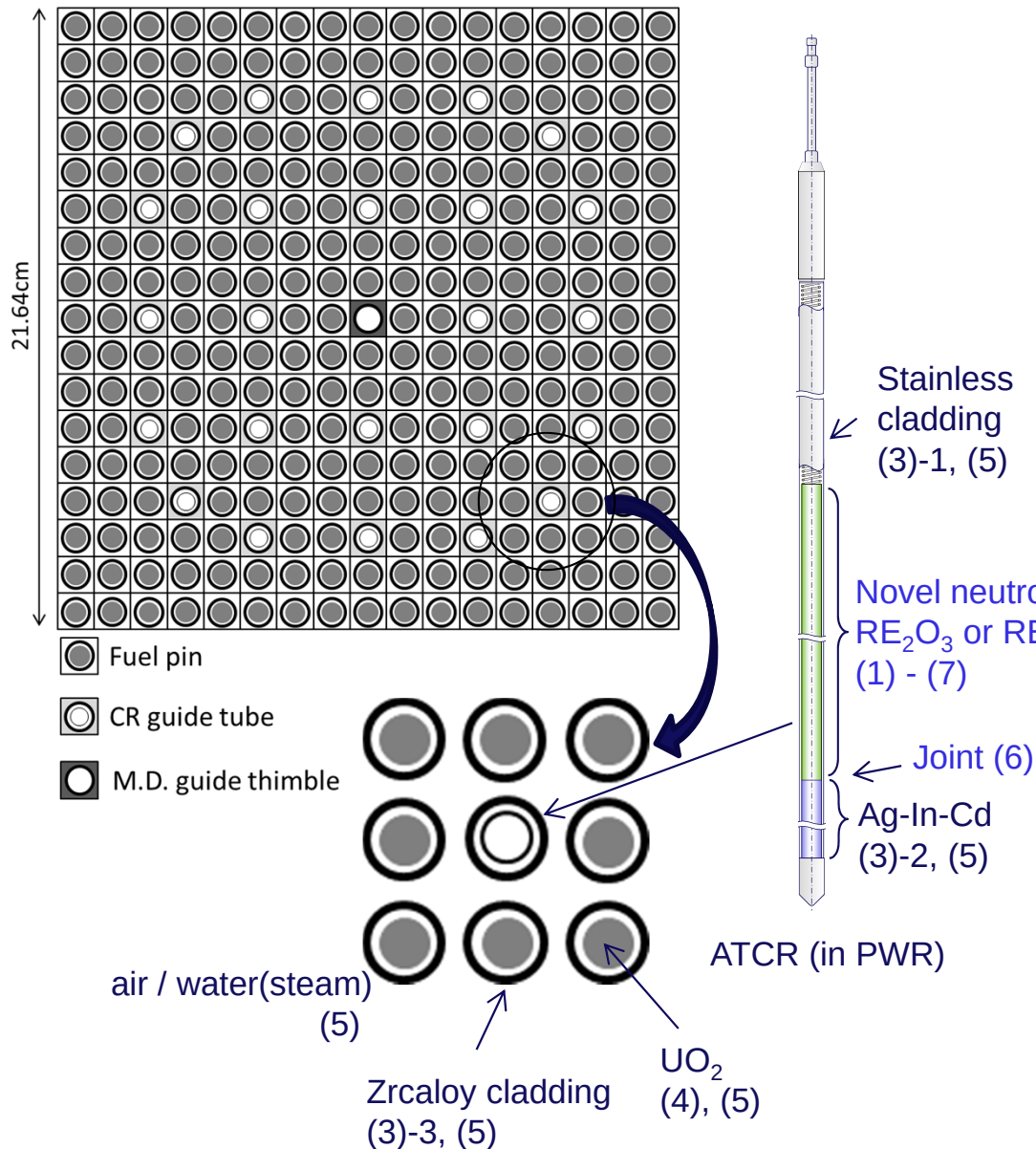
Low neutron leakage region:
RE₂O₃ or RE₂O₃-HfO₂
→ Minimum loss of neutron absorber

High neutron leakage region:
Ag-In-Cd
→ No influence on operating conditions

High neutron absorption
→ High shutdown margin

High melting or eutectic temperature
Miscibility with fuel materials
→ Remaining in the core material

Technical Issues & Development Plan – Absorber Modification –



- (1) **Reactivity worth** analysis
- (2) **Lifetime** evaluation of ATCR,
- (3) High-temperature **compatibility** observation **with structural materials**,
 1. RE_2O_3 - Fe,
 2. RE_2O_3 - AlC (or B_4C) - Fe,
 3. RE_2O_3 - Zr ...
- (4) Confirmation of **miscibility with fuel** materials, $RE_2O_3-UO_2$
- (5) **SA** condition experiment, UO_2 - Zr(Zry) - Fe(S.S.) - RE_2O_3 etc. in air / water(steam)
- (6) **Structure design & fabrication** RE_2O_3 / Ag-In-Cd(B_4C)
- (7) **Irradiation experiment**

Development Schedule of ATCR – Absorber Modification –

JFY	2014	2015	2016	2017	2018	2019	2020
(1) ATCR Worth Analysis			■ ■ ■ ■ ■ ■ ■ ■				
(2) Lifetime Evaluation							
(3) Material Tests w/o U							
(4) Material Tests with U							
(5) S.A. Condition Experiment							
(6) ATCR System Exam. & ATCR Fabrication							
(7) Irradiation Experiment & PIEs							



 ATF development project is discussed.

High temperature stability of ATCR

Experimental apparatus

High temperature furnace

: Induction heating

Available Temp. $> 2700^{\circ}\text{C}$

Atmosphere = Vacuum, Inert gases

Compatibility of Gd_2O_3 -Fe

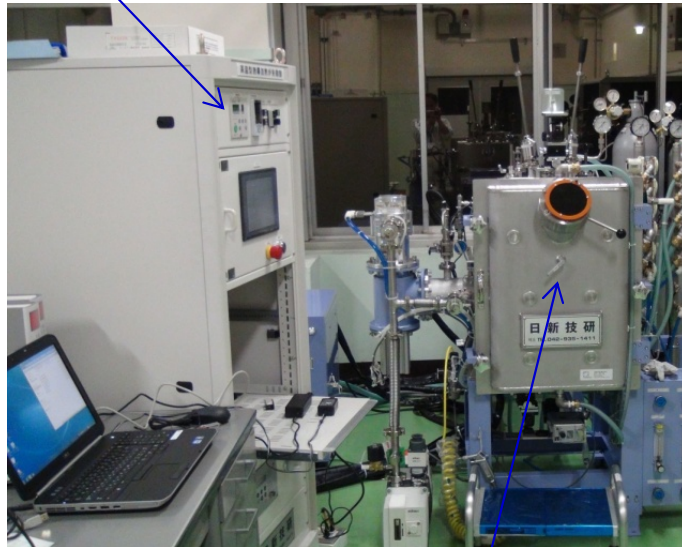
Heating of Gd_2O_3 crucible containing Fe

Temp. = 1600°C ($>$ m.p. of Fe)

Holding time = 1 hr.

Atmosphere = Ar

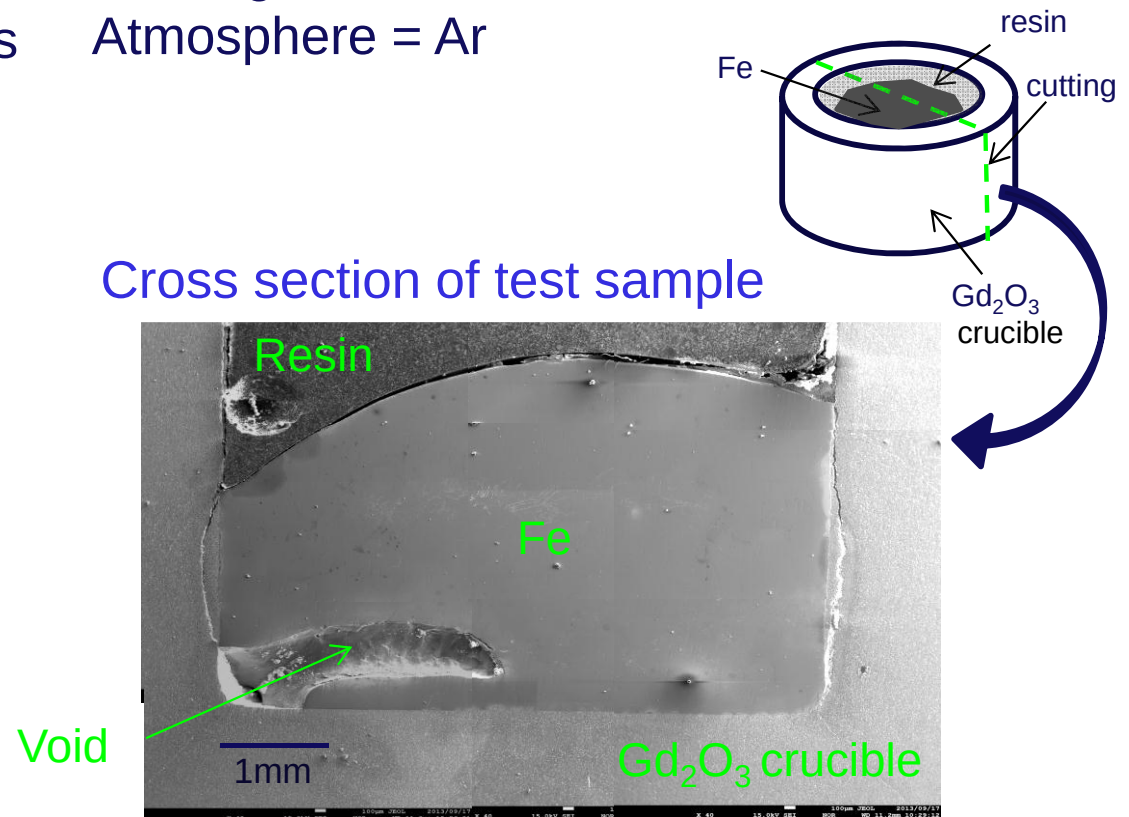
Control system



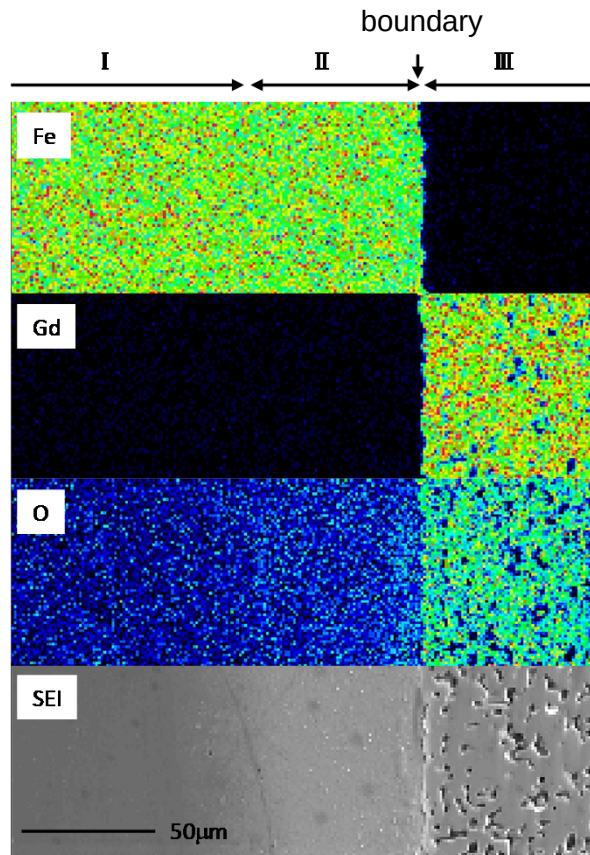
Record PC

Furnace chamber

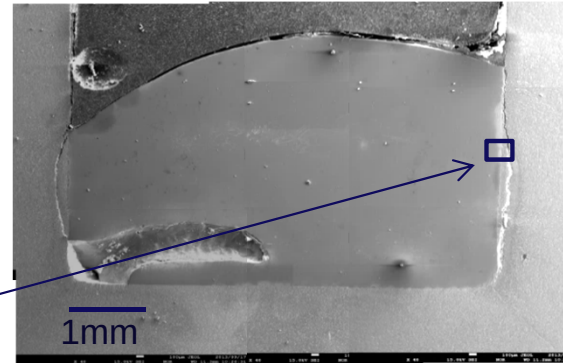
Cross section of test sample



Reaction between molten Fe and Gd_2O_3



Distributions of Fe, Gd and O around Fe/ Gd_2O_3 reaction boundary



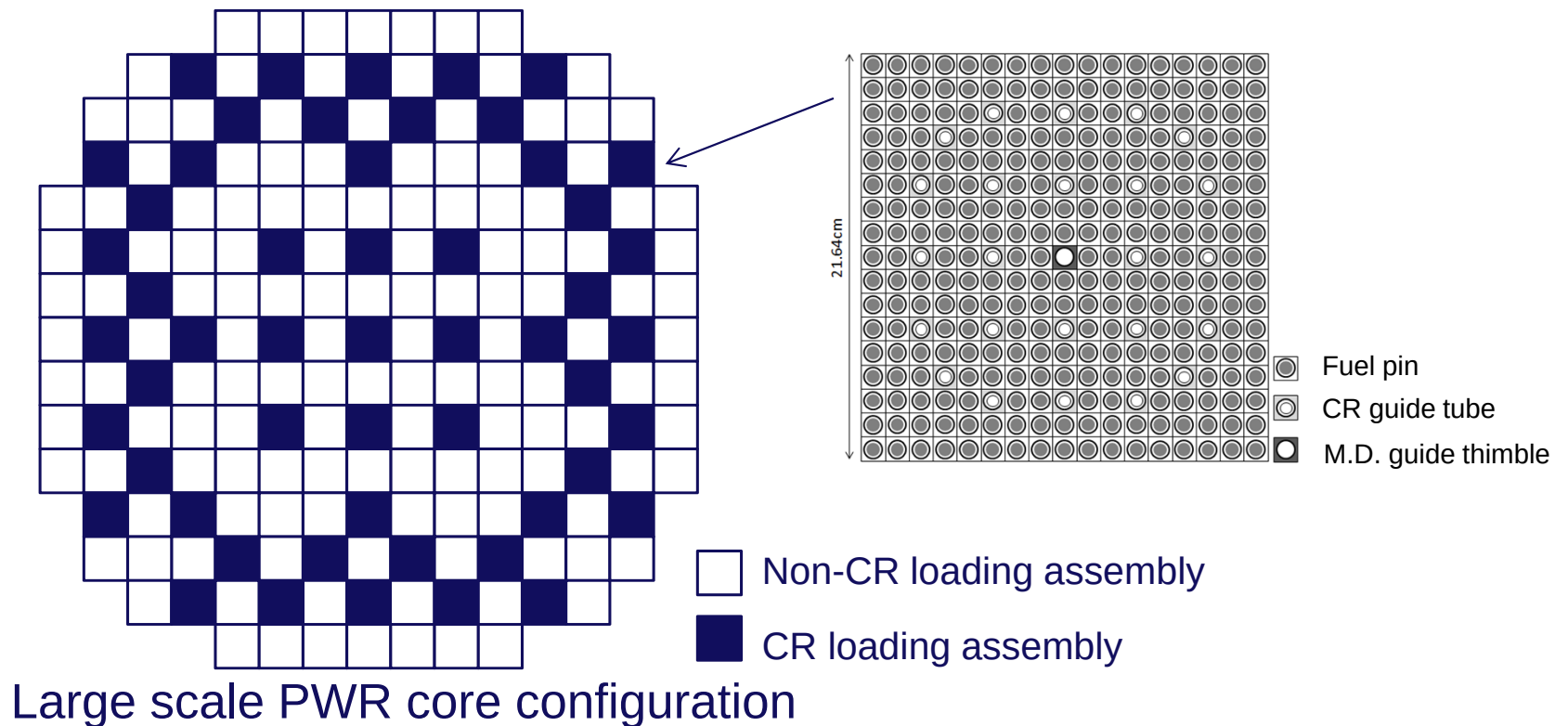
Preliminary composition analysis results [mol%]

Area	Fe	Gd	O
I	98	N.D.	2
II	93	N.D.	7
III	N.D.	47	53

← boundary

- No significant reaction was found between molten Fe and Gd_2O_3 at 1600°C.
- Slight migration of oxygen into periphery of Fe was observed.

Preliminary analysis of CR worth



CR worth analysis using MVP code

	Sm_2O_3	Eu_2O_3	Gd_2O_3	HfO_2	AIC
Reactivity worth of CR [% $\Delta k/k'$]	5.2	6.3	5.1	4.6	5.1

$$\text{Eu}_2\text{O}_3 > \text{Sm}_2\text{O}_3 \cong \text{AIC} \cong \text{Gd}_2\text{O}_3 > \text{HfO}_2$$

Summary

1. ATCR-concept proposed from CRIEPI, Japan.

- Sub-criticality attained at any accidental conditions including SA.

Candidate Materials

Absorber: RE_2O_3 (Sm_2O_3 , Gd_2O_3 , Eu_2O_3 ), HfO_2 , etc.

Structure: Fe-Cr-Al ODS ferritic steel, etc.

- Applicability to current LWRs is promising.
Influences on normal operation conditions potentially kept minimized.
CR worth potentially kept maintained for a long term.

2. R&D of ATCR were launched.

- (1) Material tests initiated.

High-temperature compatibility between Gd_2O_3 and Fe was confirmed to 1600°C.

- (2) An available CR reactivity preliminarily analyzed for candidate absorber materials.

$\text{Eu}_2\text{O}_3 > \text{Sm}_2\text{O}_3 \cong \text{Al}_2\text{O}_3 \cong \text{Gd}_2\text{O}_3 > \text{HfO}_2$

- (3) Details about structure design, fabrication & irradiation experiments being discussed.
- (4) Development of alternative structural materials (Fe-Cr-Al ODS, etc.) has started by collaborating industries.

⇒ Details will be presented in the near future.

3. Outlook of ATCR development

- ATF development project for application of current LWRs start being discussed.
- CRIEPI wants to introduce ATCR in ~ 2020.